

Diptikanta Swain and
T. N. Guru Row*Solid State and Structural Chemistry Unit, Indian
Institute of Science, Bangalore 560 012,
Karnataka, IndiaCorrespondence e-mail:
sctng@sscu.iisc.ernet.in**Key indicators**Single-crystal X-ray study
 $T = 294$ K
Mean $\sigma(S-O) = 0.002$ Å
 R factor = 0.013
 wR factor = 0.038
Data-to-parameter ratio = 14.8For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.**Rb₂Mn₂(SO₄)₃, a new member of the
langbeinite family**

The structure of a new langbeinite, dirubidium dimanganese trisulfate, Rb₂Mn₂(SO₄)₃, has been determined and is shown to have cubic symmetry akin to the other members of the langbeinite family. The structure contains SO₄ tetrahedra corner-linked to MnO₆ octahedra, generating a three-dimensional network. The Rb atoms are found in the cavities of this network.

Received 12 May 2006

Accepted 24 May 2006

Comment

Rb₂Mn₂(SO₄)₃ belongs to the langbeinite family, having the general formula $A_2B_2(XO_4)_3$, where $A = \text{Rb, Li, NH}_4, \text{ Tl, K or Cs}$, $B = \text{Mg, Ca, Mn, Fe, Co, Ni, Cu, Zn, Cd}$ with $X = \text{S, Se or Mo}$. The name originates from the natural langbeinite K₂Mg₂(SO₄)₃, the crystal structure of which was first solved by Zemmann & Zemmann (1957). A new type of langbeinite was synthesized by replacing the SO₄ group by a BeF₄ group (Gueylyah *et al.*, 1996), which exhibited several interesting low-temperature phase transitions. Langbenites are well known to exhibit ferroelectric and ferroelastic phase transitions with temperature. There are three types of langbenites; type-I has a room-temperature cubic structure and undergoes several phase transitions to a final orthorhombic crystal form (space group $P2_12_12_1$) via a monoclinic crystal system (space group $P2_1$) followed by a triclinic crystal system (space group $P1$) with lowering of temperature. Type-II langbenites show a single phase transition when cooled, resulting in an orthorhombic crystal form (space group $P2_12_12_1$). Type-III langbeinite remains invariant with lowering of temperature.

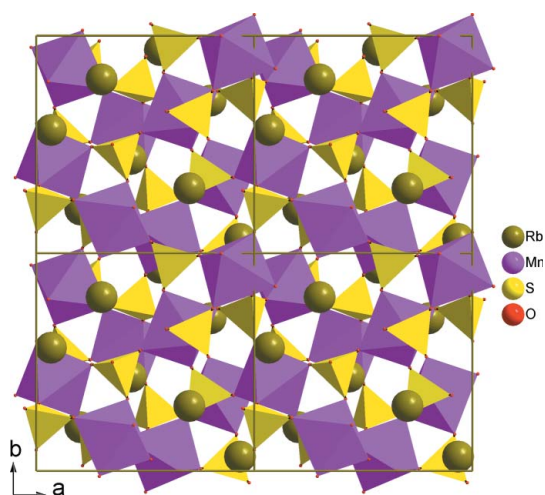


Figure 1
Packing diagram, viewed down the c axis, depicting the three-dimensional network of MnO₆ octahedra and SO₄ tetrahedra with Rb atoms in the cavities.

However, in a recent study Nalini & Guru Row (2002) have shown that $\text{Rb}_2\text{Cd}_2(\text{SO}_4)_3$ shows no intermediate triclinic form in Type-I. The title compound is isostructural with the rest of the known langbeinite family, with MnO_6 octahedra corner-shared with the SO_4 tetrahedra, as shown in Fig. 1, forming a three-dimensional network. Rb2 is nine-coordinate, whereas Rb1 is 12-coordinate, with both atoms residing in the cavities.

Experimental

The title compound was synthesized by slow evaporation at 393 K of an aqueous solution containing equimolar amounts of Rb_2SO_4 and MnSO_4 . The evaporation rate was slowed down considerably by sealing the 5 ml beakers containing the solution in a thermostat.

Crystal data

$\text{Rb}_2\text{Mn}_2(\text{SO}_4)_3$	$D_x = 3.547 \text{ Mg m}^{-3}$
$M_r = 569.03$	Mo $K\alpha$ radiation
Cubic, $P2_13$	$\mu = 12.10 \text{ mm}^{-1}$
$a = 10.2140 (7) \text{ \AA}$	$T = 294 (2) \text{ K}$
$V = 1065.58 (13) \text{ \AA}^3$	Block, colourless
$Z = 4$	$0.4 \times 0.2 \times 0.15 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	8890 measured reflections
φ and ω scans	873 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	860 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.070$, $T_{\max} = 0.160$	$R_{\text{int}} = 0.031$
	$\theta_{\max} = 28.6^\circ$

Refinement

Refinement on F^2	$(\Delta/\sigma)_{\max} = 0.001$
$R[F^2 > 2\sigma(F^2)] = 0.013$	$\Delta\rho_{\max} = 0.26 \text{ e \AA}^{-3}$
$wR(F^2) = 0.038$	$\Delta\rho_{\min} = -0.27 \text{ e \AA}^{-3}$
$S = 1.23$	Extinction correction: SHELXL97
873 reflections	Extinction coefficient: 0.0102 (5)
59 parameters	Absolute structure: Flack (1983),
$w = 1/[\sigma^2(F_o^2) + 0.4692P]$	339 Friedel pairs
where $P = (F_o^2 + 2F_c^2)/3$	Flack parameter: 0.013 (8)

Table 1

Selected bond lengths (\AA).

Rb1—O4	2.969 (2)	Mn1—O2 ⁱⁱ	2.176 (2)
Rb1—O2 ⁱ	3.122 (2)	Mn2—O4 ^{vi}	2.150 (2)
Rb1—O1	3.229 (2)	Mn2—O1 ^{vii}	2.173 (2)
Rb1—O1 ⁱⁱ	3.257 (2)	S1—O4	1.463 (2)
Rb2—O3 ⁱⁱⁱ	2.948 (2)	S1—O2	1.466 (2)
Rb2—O2 ^{iv}	3.087 (2)	S1—O1	1.475 (2)
Rb2—O1 ^{iv}	3.196 (2)	S1—O3	1.477 (2)
Mn1—O3 ^v	2.155 (2)		

Symmetry codes: (i) $z, x, y - 1$; (ii) $y - \frac{1}{2}, -z + \frac{3}{2}, -x + 1$; (iii) $x - 1, y, z$; (iv) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$; (v) $-z + \frac{1}{2}, -x + 2, y - \frac{3}{2}$; (vi) $x - \frac{1}{2}, -y + \frac{3}{2}, -z$; (vii) $-y + \frac{3}{2}, -z + 1, x - \frac{1}{2}$.

Data collection: SMART (Bruker, 2004); cell refinement: SMART; data reduction: SAINT (Bruker, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: DIAMOND (Brandenburg, 1999); software used to prepare material for publication: PLATON (Spek, 2003).

The authors thank the Department of Science and Technology, India (IRHPA–DST), for providing the CCD facility at the Indian Institute of Science.

References

- Brandenburg, K. (1999). *DIAMOND*. Version 2.1c. Crystal Impact GbR, Bonn, Germany.
- Bruker (2004). *SMART* (Version 5.628) and *SAINTE* (Version 6.45a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Guelylah, A., Breczewski, T. & Madariaga, G. (1996). *Acta Cryst.* **C52**, 2951–2954.
- Nalini, G. & Guru Row, T. N. (2002). *Chem. Mater.* **14**, 4729–4735.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Zemann, A. & Zemann, J. (1957). *Acta Cryst.* **10**, 409–413.